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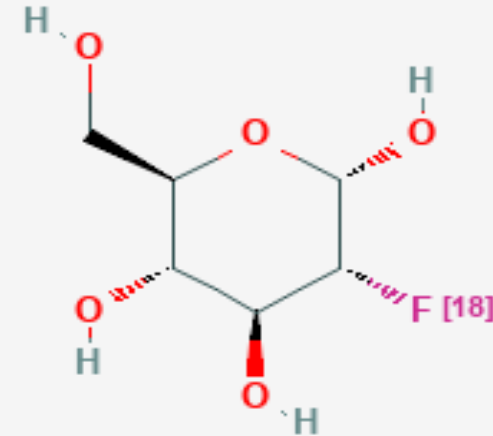
# Characteristics and Applications of Non-Conventional Radionuclides in Radiopharmaceuticals for Medical Imaging

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STUDENT: NATIA INADZE

# What are Radiopharmaceuticals and Radioisotopes?

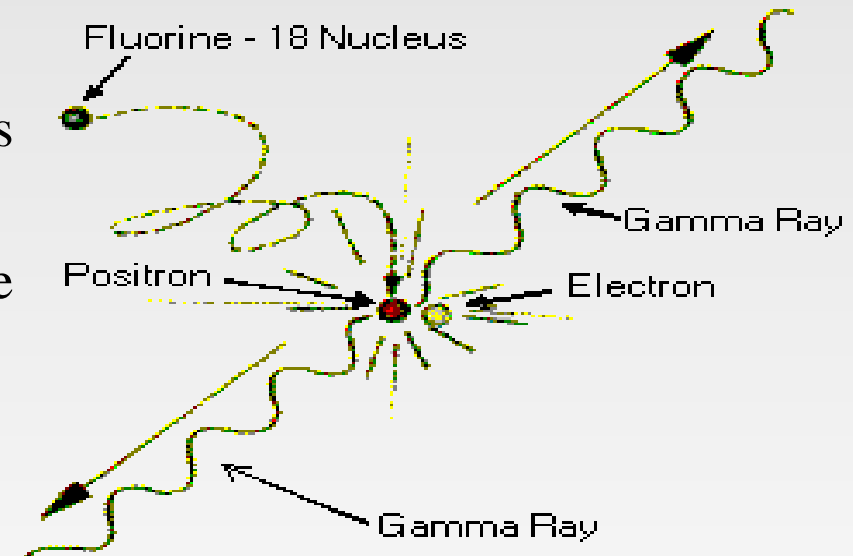
- Radiopharmaceuticals are a group of diagnostic or therapeutic agents synthesized by linking various radionuclides to biological molecules
- Radiopharmaceuticals labelled with b+-emitting radionuclides are utilized in medical imaging and are detected by their annihilation gamma quanta
- Radioisotopes are unstable isotopes, spontaneously dissipating excess energy in the form of radiation to produce a nonradioactive daughter atom
- Most common radionuclides used for diagnostic purposes are  $^{99}\text{Tc}$  ( $T_{1/2} = 6 \text{ h}$ ) frequently used in SPECT,  $^{18}\text{F}$  ( $T_{1/2} = 110\text{min}$ ),  $^{11}\text{C}$  ( $T_{1/2} = 20.4\text{min}$ ),  $^{13}\text{N}$  ( $T_{1/2} = 9.96\text{min}$ ),  $^{15}\text{O}$  ( $T_{1/2} = 2.04\text{min}$ ), all used in PET



# Diagnostic Radioisotope Characteristics

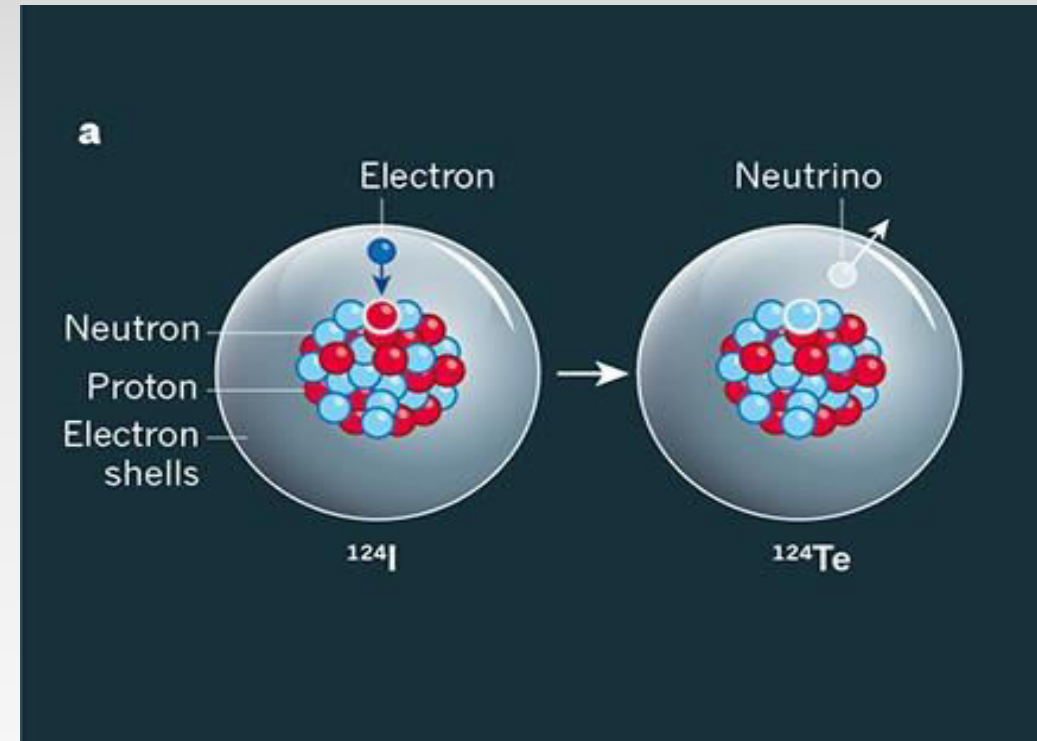
- Their half-life should correspond to the residence time of the radiopharmaceutical in the organism
- The yield of the positrons should be maximum, and, correspondingly, the positron energy should be as low as possible
- The presence of high-energy gamma lines in the nuclide spectrum should be avoided
- The radioisotope should form kinetically and thermodynamically stable compounds *in vivo*

## Positron Emission Tomography



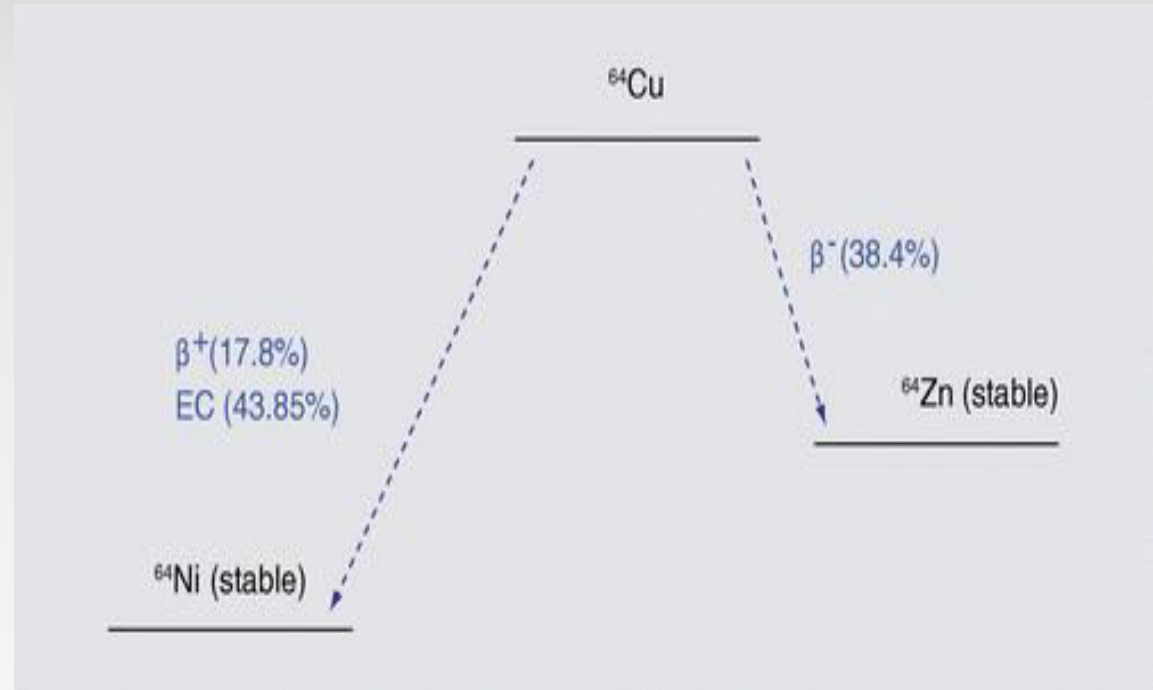
# Rise of Non-Conventional Radionuclides

- For the purpose of studying slower functional processes, involving protein synthesis, cell proliferation, etc., nonstandard, longer-lived positron emitters are needed
- Some of them include  $^{64}\text{Cu}$  ( $T_{1/2} = 12.7$  h);  $^{86}\text{Y}$  ( $T_{1/2} = 14.7$  h),  $^{89}\text{Zr}$  ( $T_{1/2} = 3.3$  d),  $^{124}\text{I}$  ( $T_{1/2} = 4.2$  d),  $^{73}\text{Se}$  ( $T_{1/2} = 7.1$  h), and many others



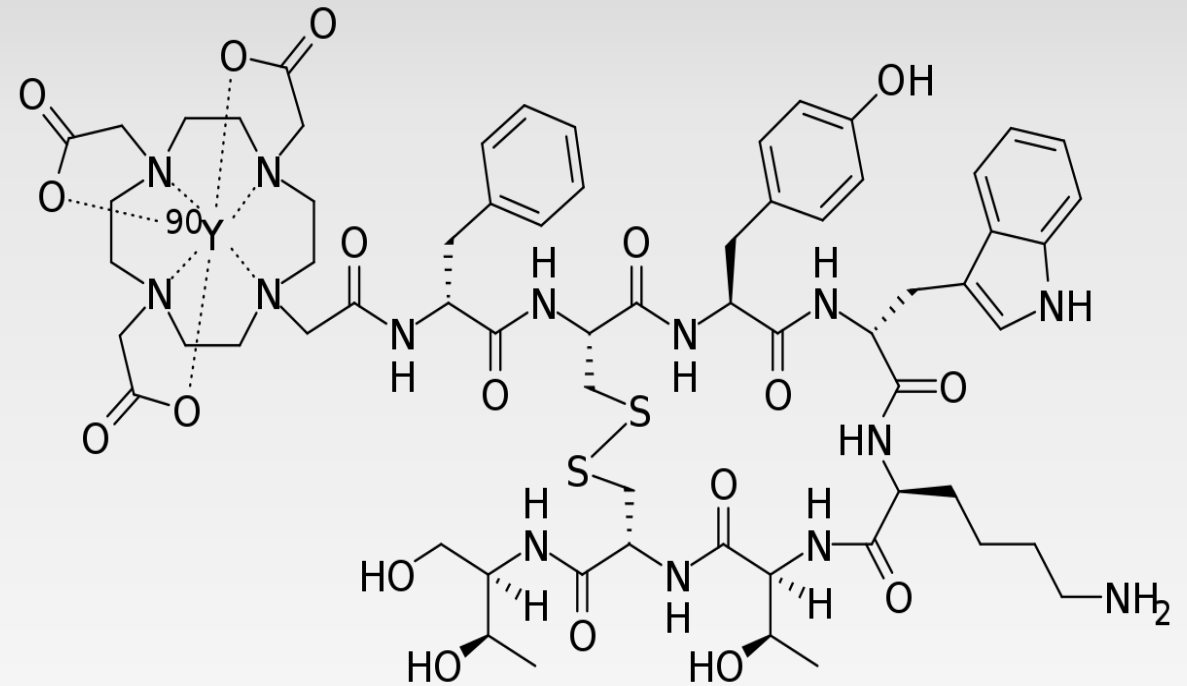
# Copper-64

- The most commonly used non-conventional PET radioisotope
- The physical decay properties of  $^{64}\text{Cu}$  include 18%  $\beta^+$ -, 38%  $\beta^-$ - emission and 44% by electron capture
- Relatively low  $\beta^+$  energy of 0.653 MeV
- Cu-64 forms quite stable coordination complexes *in-vivo*



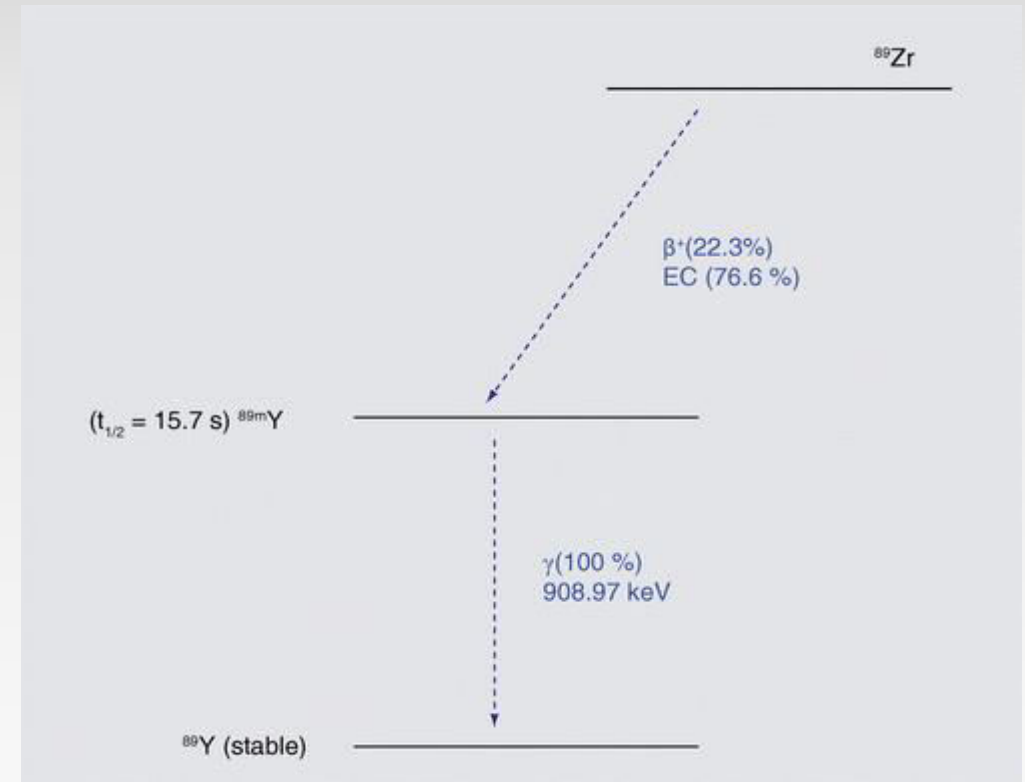
# Yttrium- 86

- Widely known metallic non-standard PET radionuclide
- $^{86}\text{Y}$  has a half-life of 14.7 h and decays by  $\beta^+$ - (34%) and  $\gamma$ - (77%) emission
- It has gained attention as an imaging pair for  $^{90}\text{Y}$ , a therapeutic radionuclide currently used in nuclear medicine.
- The limitation of this  $\beta^+$  isotope is relatively high positron energy of 3.15MeV and the presence of additional  $\gamma$ -rays that are emitted during  $^{86}\text{Y}$  decay.



# Zirconium-89

- It is a long-lived radionuclide with a half-life of 78.4 h and decays by electron capture,  $\beta^+$ -emission and  $\gamma$ -emission
- relatively low positron energy ( $E_{\beta^+} = 0.90$  MeV) results in high resolution of  $^{89}\text{Zr}$  images that are comparable to those observed with  $^{18}\text{F}$  and  $^{64}\text{Cu}$



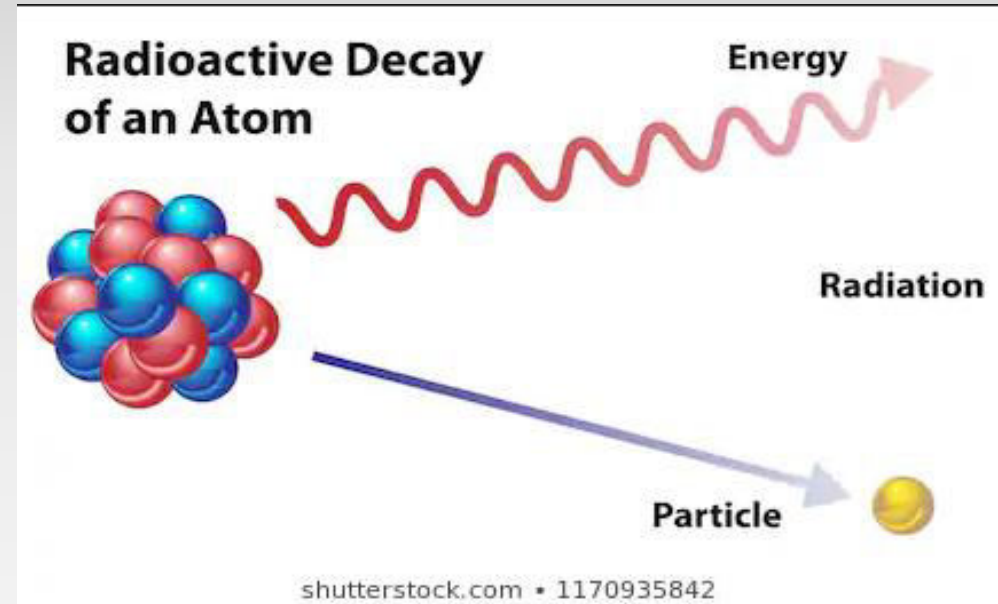
# Other Non-Standard Radionuclides

## IODINE-124

- Is somewhat longer-lived ( $T_{1/2} = 4.18\text{d}$ ), and has relatively low positron branching of 22.0%.
- The  $\beta^+$  endpoint energy amounts to 2.13 MeV, and several  $\gamma$ -rays are also associated with its decay.

## SELENIUM-73

- It decays 65.4% via  $\beta^+$  emission and 34.6% by EC. The positron endpoint energy amounts to 1.3 MeV and a few  $\gamma$ -rays are also emitted.

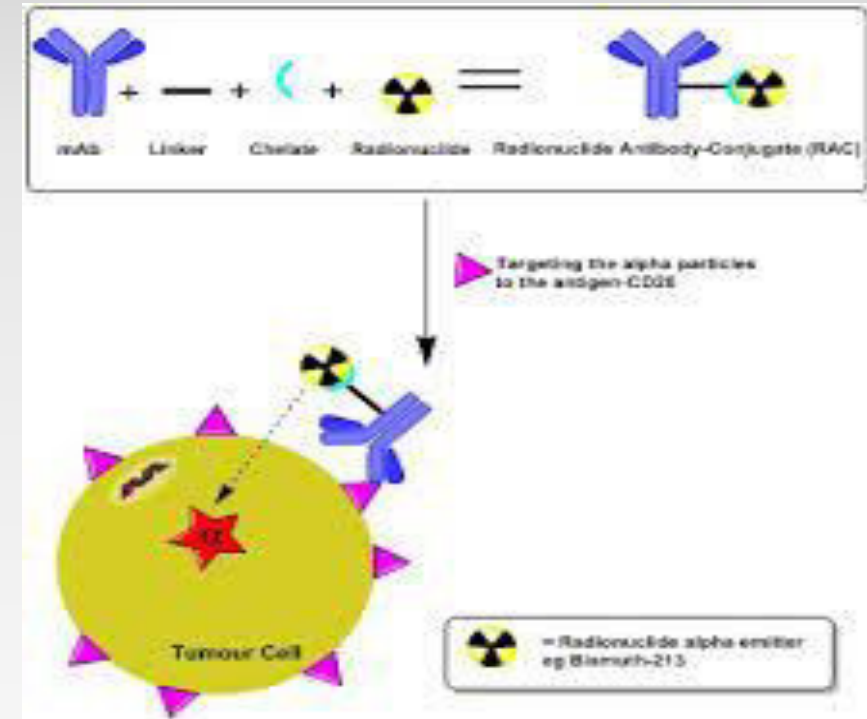


Many other non-traditional radionuclides are also known [ $^{43}\text{Sc}$  ( $T_{1/2} = 3.9\text{ h}$ ),  $^{44}\text{Sc}$  ( $T_{1/2} = 3.9\text{ h}$ ),  $^{45}\text{Ti}$  ( $T_{1/2} = 3.1\text{ h}$ ),  $^{52}\text{Mn}$  ( $T_{1/2} = 5.6\text{ d}$ ),  $^{55}\text{Co}$  ( $T_{1/2} = 17.5\text{ h}$ ),  $^{61}\text{Cu}$  ( $T_{1/2} = 3.3\text{ h}$ ),  $^{120}\text{I}$  ( $T_{1/2} = 1.3\text{ h}$ ), etc.]



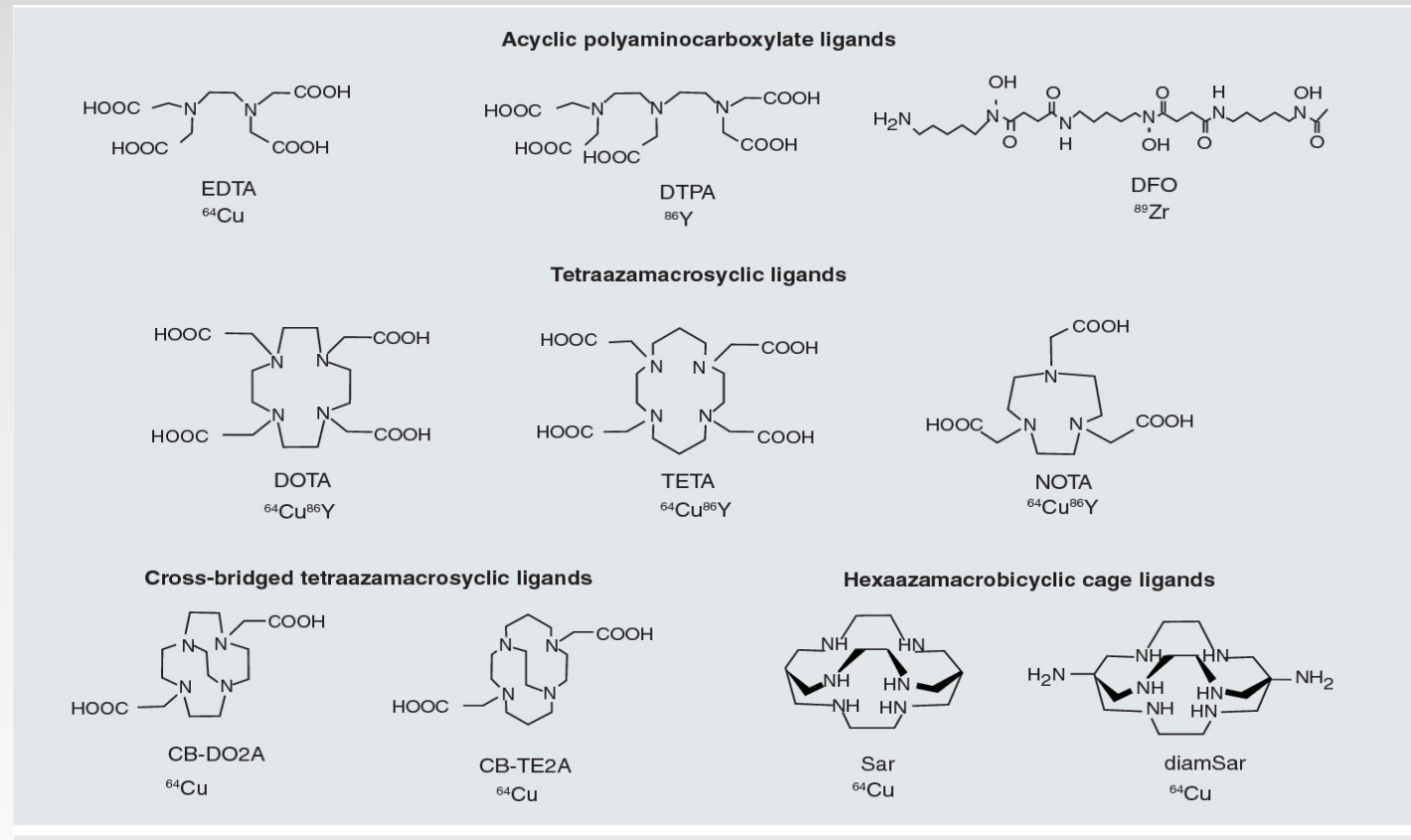
# Radiotracer Preparation - Chelators

- Incorporation of longer-lived metal isotopes is based on binding a radionuclide with a bifunctional chelating agent (BCA) that, in turn, can form covalent bonds with various biomolecules (vectors), with high specificity and selectivity
- Radiopharmaceuticals based on metal isotopes are not involved in cellular metabolism but bind to the target, for example, by interactions of the vector molecule with receptors expressed on the tumor surface.



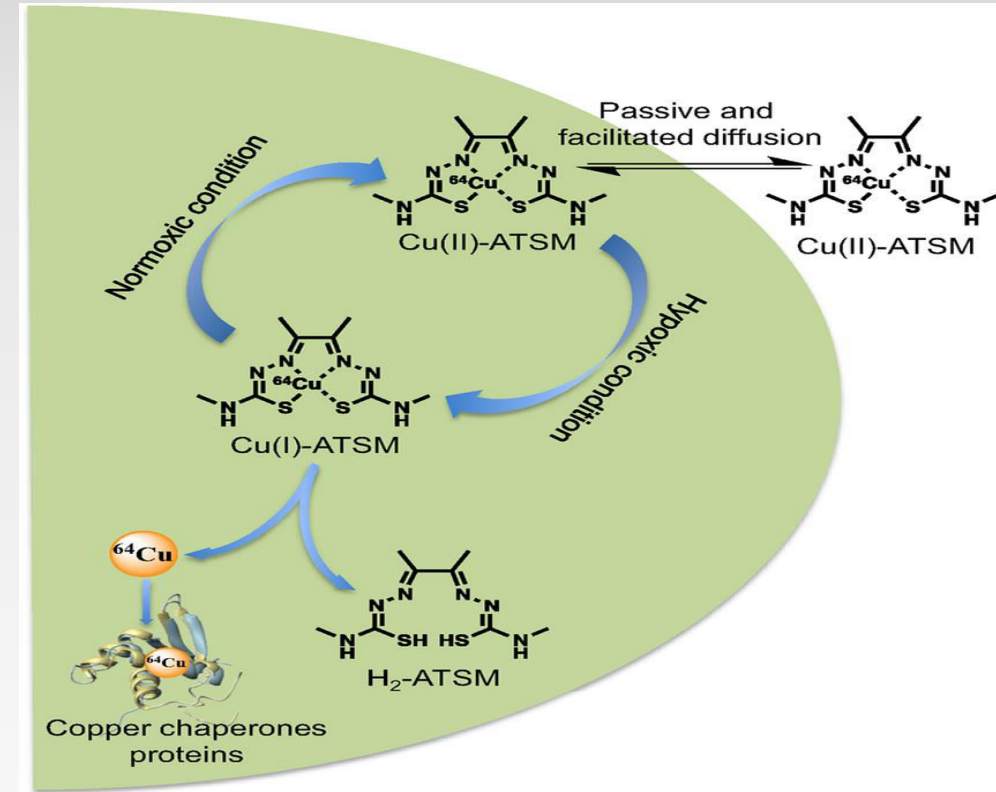
# Bifunctional Chelators (BCAs)

- The macrocyclic polyaminocarboxylates, 1,4,7,10-tetraazacyclododecanetetraacetic acid (DOTA) and 1,4,8,11-tetraazacyclotetradecanetetraacetic acid (TETA), as well as their derivatives, are the most frequently used type of chelators, satisfying the necessary conditions for BCAs



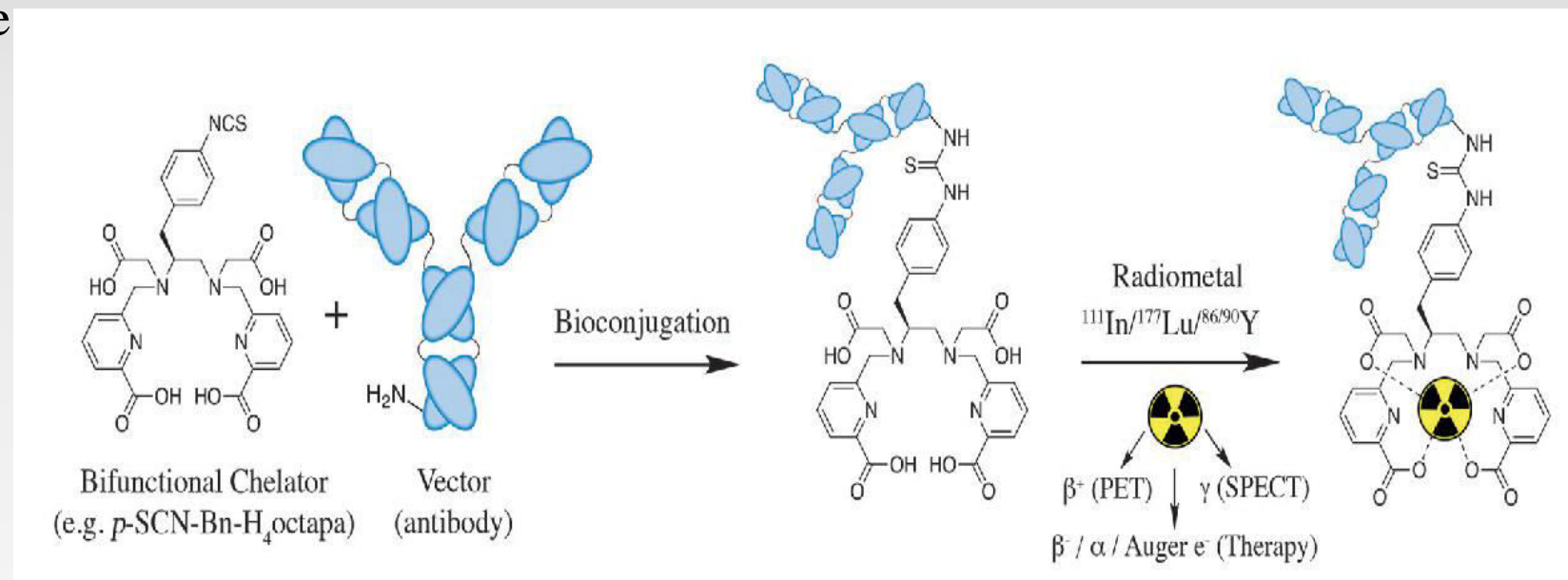
# Small-Molecule Probes

- One group of special interest is the semicarbazone ligand that has been complexed with  $^{64}\text{Cu}$  and investigated to probe hypoxia
- With the semicarbazone ligand, diacetyl-*bis*( $\text{N}^4$ -methylthiosemicarbazone) (ATSM), which has a maximum positron energy of 0.66 MeV, similar to that of  $^{18}\text{F}$ , the resulting PET images are the best obtainable with any positron-emitting radionuclide of copper.



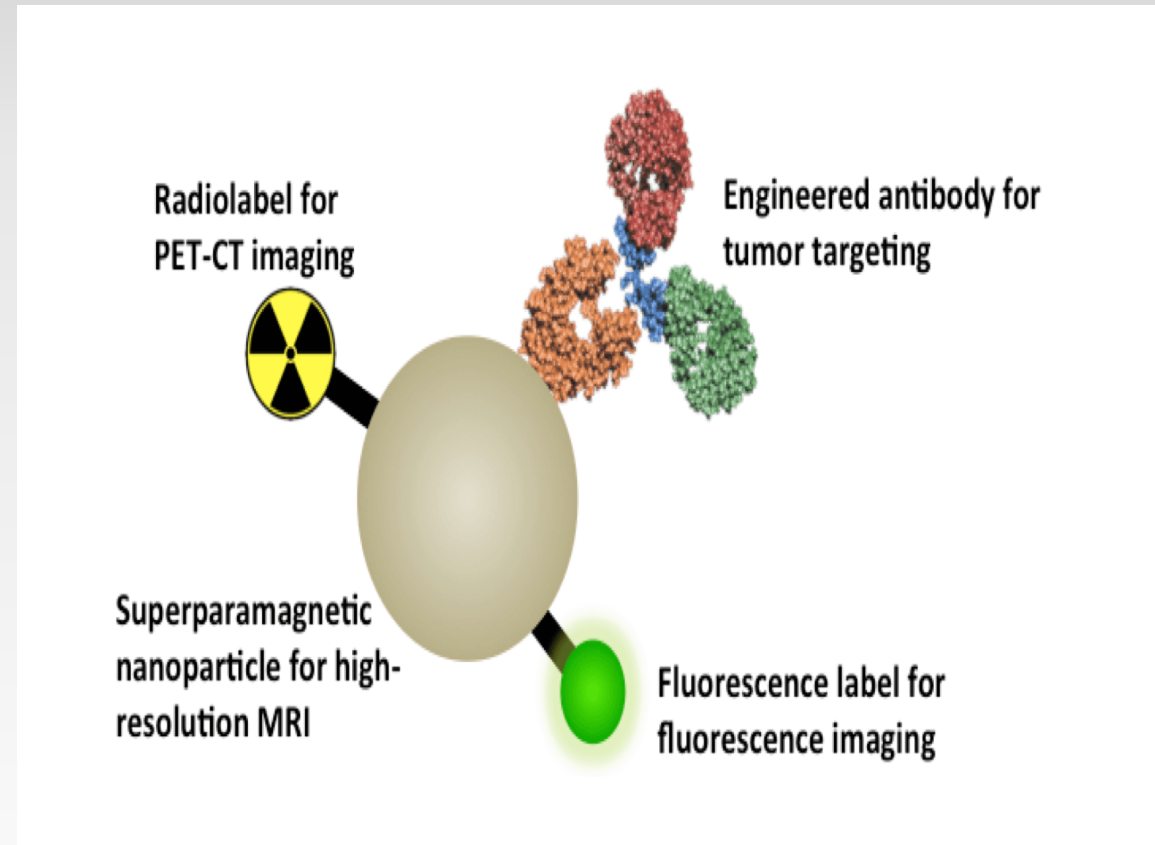
# Peptide Probes

Peptide probes utilize the potential targets, such as somatostatin (SST) receptors, integrins and gastrin-releasing peptide receptors (GRPRs), which are overexpressed in many cancer cell types for imaging purposes.



# Nanoparticle Probes

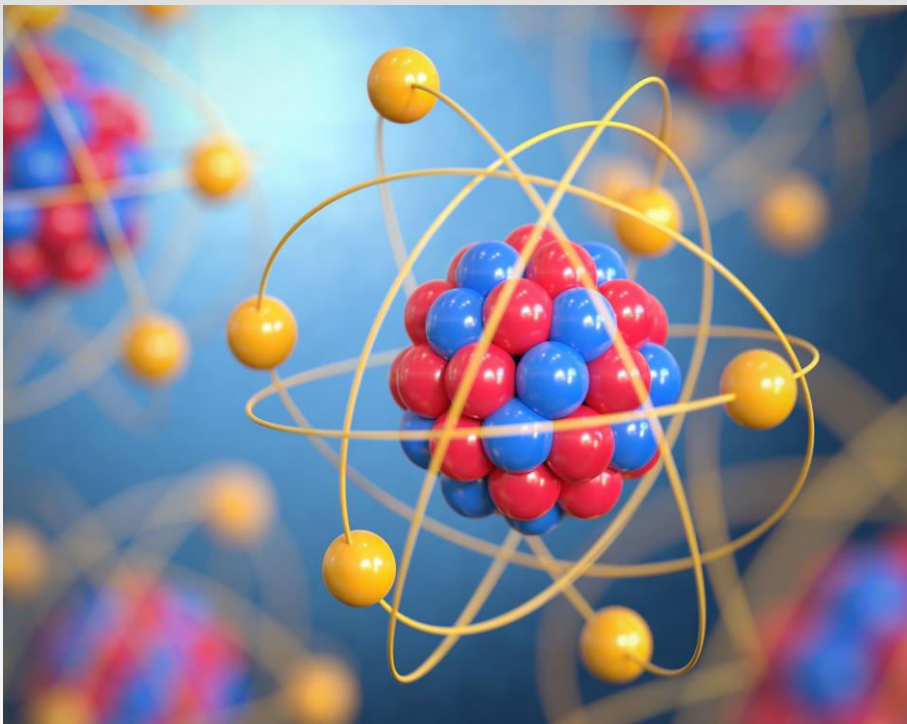
- Nanoparticle probes are produced as diagnostic agents primarily because of their small size, large surface area and biokinetics.
- These properties permit greater functionality and wider *in vivo* distribution than can be accomplished with other molecular probes.
- A single nanoparticle can be conjugated with a large number of reporter molecules effectively, such as radionuclides or fluorophores, increasing signal-to-noise ratios.





# Summary

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- The nonstandard isotopes play a key role in molecular imaging probe design utilized for studying slow metabolic process.
- The long half-lives and chemical versatility of these radionuclides have fueled novel probe development and enabled imaging of various biological processes.
- These isotopes have been effective in oncologic, cardiovascular and neurological imaging.
- Future research areas include further development of the strategies for radiopharmaceutical synthesis and production to avoid problems associated with the non-standard nuclides, such as non-specific tissue accumulation for example, as well as to increase the availability of these radioisotope radiopharmaceuticals in wider-scale medical applications.

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**Thank You for Attention!**

