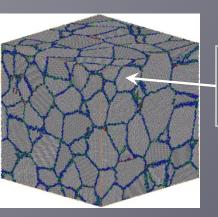
### Development and Investigation of Nanocrystalline Composite Materials

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#### Nanocrystalline Materials

Nanocrystalline material- is any shaped bulk polycrystalline sample with grain size till 100 nm in at least one dimension



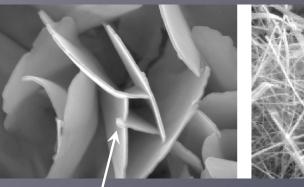
1 Dimensional

Nanomaterial

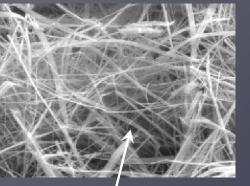
3D Grain no mare 100 nm

The main techniques for manufacturing nanopowders: synthesis by solid-state reaction; spray pyrolysis, pulsed laser deposition thermal synthesis; synthesis in salt melts; hydrothermal synthesis; sol-gel synthesis.

Standard methods for manufacturing of bulk material; cold compaction with further sintering; hot pressing; sintering under high pressure; electric discharge synt.; shock-wave sintering ; gasostate sintering; hot isostatic pressing; Spark Plasma Synthesis (SPS).



NbSe<sub>2</sub>



2 Dimensional Nanomaterial The most universal method for manufacturing ; Nanopowders - <u>sol-gel synthesis</u> Bulk material - <u>Spark Plasma</u> <u>Synthesis</u>



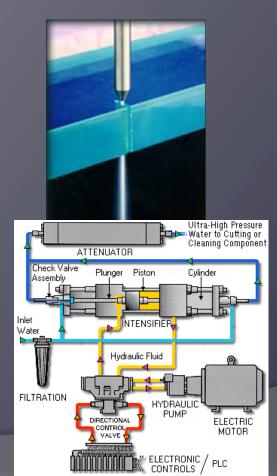
#### Materials we are developing

Scintillating materials: *Silicates, LSO, YSO, Aluminates , LuAP, LuAG Tungstats: PWO,CdWO*<sub>4,</sub> CaWO<sub>4,</sub>

Hard Metals: WC-Co TiC-Ni-Mo-W TiC-Fe-Ni

Armor materials:  $B_4C$   $B_4C$ -Cu-Mn TiC-Ni-Mo-W  $TiB_2$  - TiC









#### Scintillators

<u>Standart:</u>

Silicates - Lu<sub>2</sub>SiO<sub>5</sub>:Ce (LSO), Lu<sub>2</sub>Si<sub>2</sub>O<sub>7</sub>:Ce (LPS), Y<sub>2</sub>SiO<sub>5</sub>:Ce (YSO) Aluminates - LuAlO<sub>3</sub>:Ce (LuAP), Lu<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:Ce (LuAG), Y<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:Ce (YAG) Tungstates - PbWO<sub>4</sub> (PWO), ZnWO<sub>4</sub> (ZWO), CaWO<sub>4</sub>, CdWO<sub>4</sub>, Lu<sub>2</sub>WO<sub>6</sub> <u>New type scintillators:</u>

Titanites-  $Lu_2TiO_5:Eu$ ,  $Lu_2Ti_2O_7:Eu$ , hafnates- SrHfO<sub>3</sub>:Ce, Sr<sub>2</sub>HfO<sub>4</sub>:Ce phosphates- LuPO<sub>4</sub>:Ce.

Properties of the Ideal Scintillation Crystal :

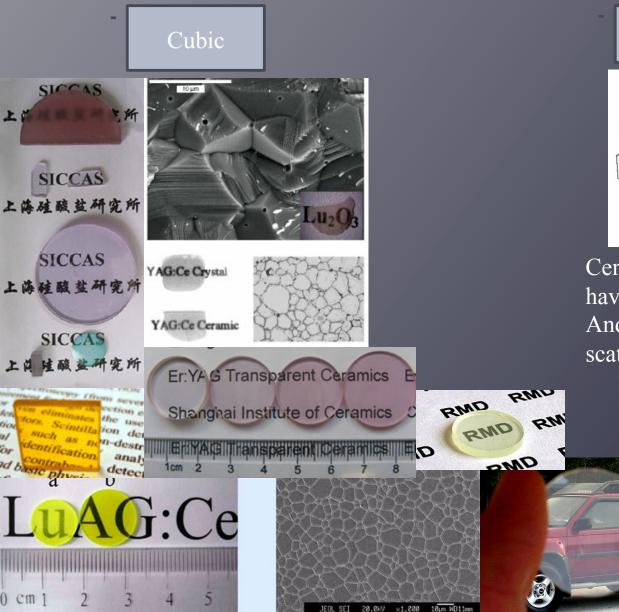
- 1. High density,
- 2. Chemical stability and mechanical strength,
- 3. High atomic number
- 4. Short decay time
- 5. High light output
- 6. Good energy resolution
- 7. Emission wavelength near 400 nm
- 8. Transparent at emission wavelength
- 9. Index of refraction near ~1.5
- 10. Non hygroscopic
- 11. Economic growth process

<u>High output</u> of Monocrystals and low obtaining cost.

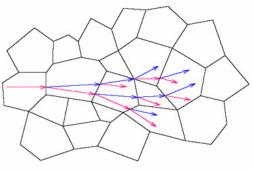


#### Scintillation ceramics as an alternative of monocrystals





#### NonCubic



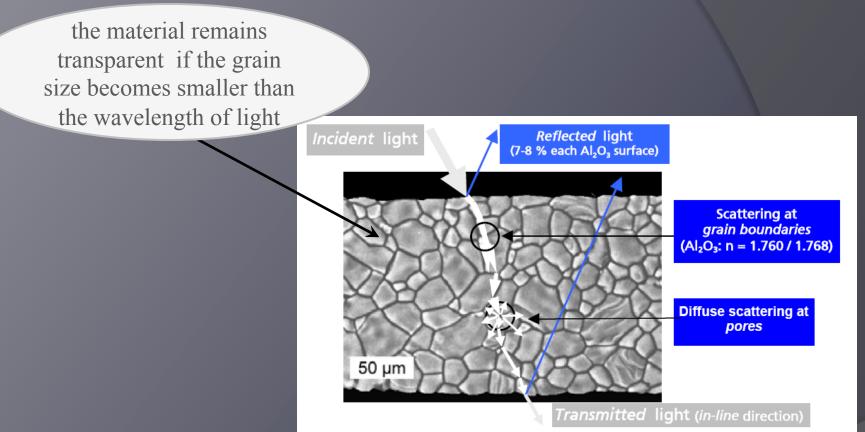
Ceramic with non cubic symmetry have multiple indices of refraction. And therefore leads to multiple scattering of the light.

TAL MAN



#### Scintillation ceramics as an alternative of monocrystals

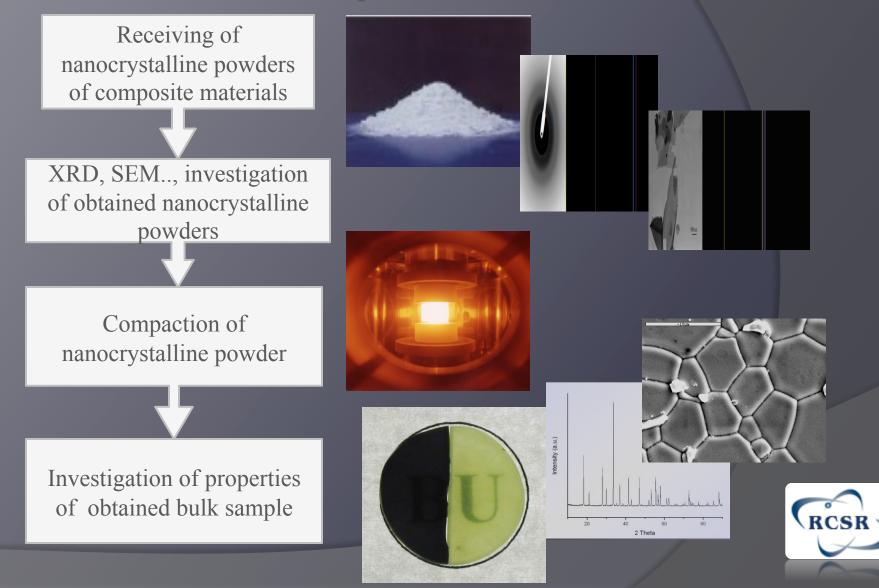




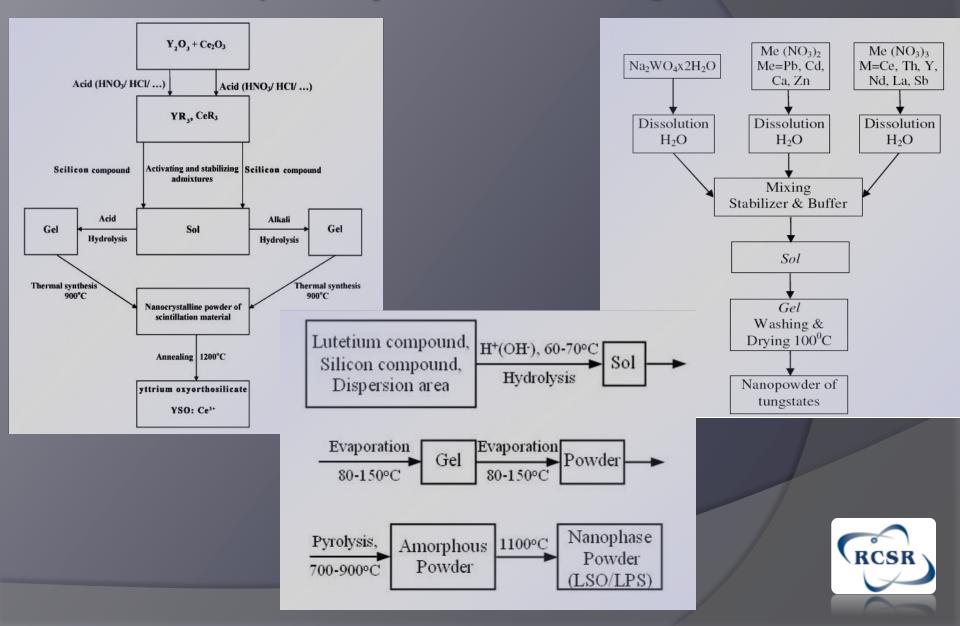
Light transmission through polycrystalline alumina (Al<sub>2</sub>O<sub>3</sub>); works of Andreas Krell, Thomas Hutzler, Jens Klimke "Physics and Technology of Transparent Ceramic Armor:"



#### Technological process of manufacturing nanocrystalline composite materials



#### General scheme of the process developed for the synthesis of nanocrystalline powders of scintillating materials



#### YSO Single Cristal Growth via nanopowder.



YSO:Ce 0.005% Single Crystal Growth via nanopowder.

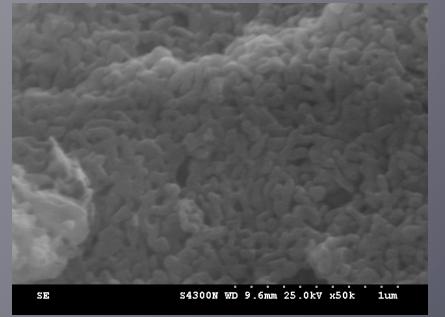
- a Remained nanopowder  $Y_2SiO_5$  after cold loading.
- b Volume comparison between
  - 1- densified Y<sub>2</sub>SiO<sub>5</sub>, 2-Standard Y<sub>2</sub>SiO<sub>5</sub>
  - 3- nanopowder  $Y_2SiO_5$
- c Powder charging comparison between
  - 1- densified Y<sub>2</sub>SiO<sub>5</sub>, 2-Standard Y<sub>2</sub>SiO<sub>5</sub>
  - 3- nanopowder  $Y_2SiO_5$

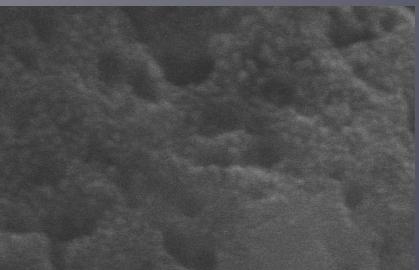


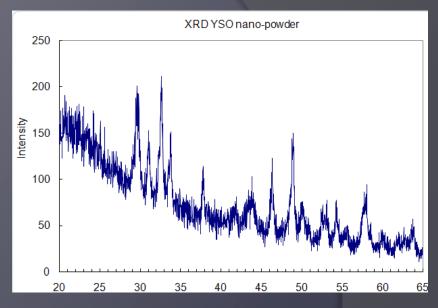
#### Growth station



#### Y<sub>2</sub>SiO<sub>5</sub> nanopowder used for crystal growth





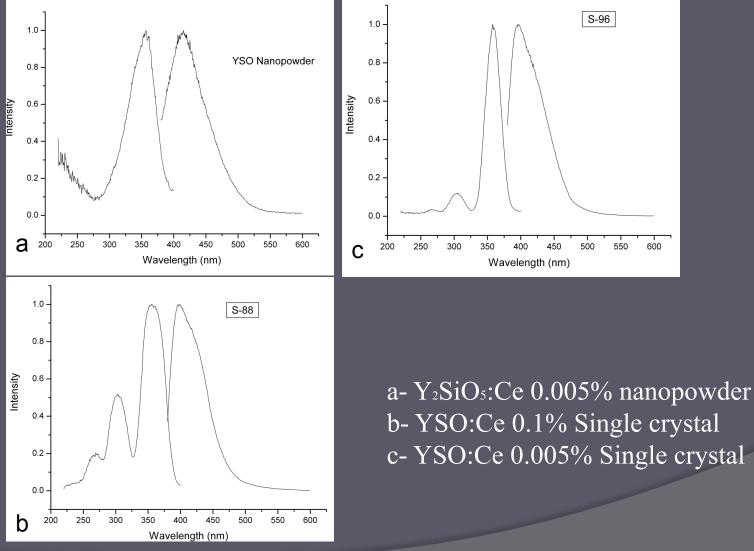


XRD of Nanopowder of Y<sub>2</sub>SiO<sub>5</sub>

Electron-microscopic images of the  $Y_2SiO_5$  nanopowder, grain size ~ 20-30 nm

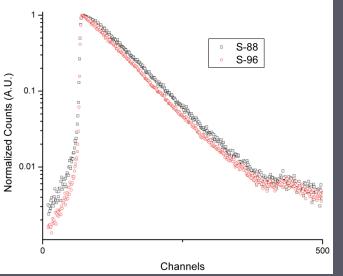


# Photoluminescence excitation and emission spectra





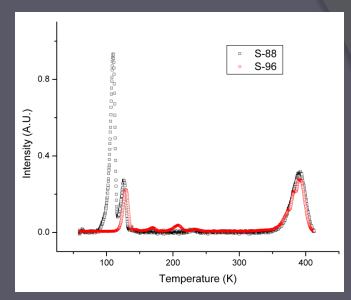
#### Scintillation properties



Decay time comparison of S-88 - YSO:Ce 0.1% Single crystal and S-96- YSO:Ce 0.005% Single crystal (received from nanopowder)

#### **Results of compression measurements**

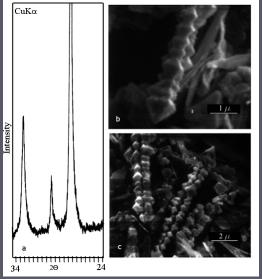
- Shorter decay time 56ns (decay time of ordinary crystal was 76ns);
- increased light yield;
- better energy resolution in respect to uniformity;
- restrain emission in the  $\sim$  300nm range of wavelength;
- increased absorption in the shorter-wavelength ;
- uniformity;
- limited percentage of the powder charge and easier loading process:

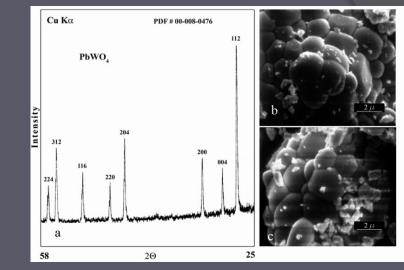


Thermoluminescence Spectrum of YSO:Ce 0.1% and YSO:Ce 0.005%

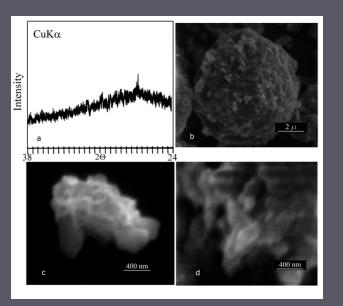


#### **PWO nanopowder**

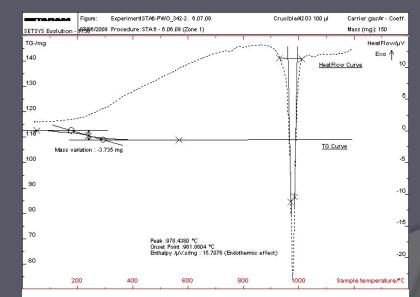




#### PWO crystalline powder: a- XRD pattern ; b,c, - SEM microg.

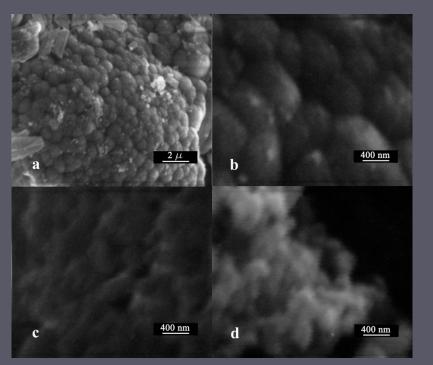


PWO amorphous for X-rays; SEM micrographs at different magnifications



TG and DTA curves of PWO nanopowder nanocrystalline state provides temperature drop of the melting point of the material almost by 155 °C, the melting point of lead tungstate being 1123°C.

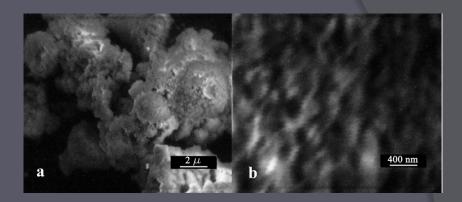
#### Nanopowder of Tungstates



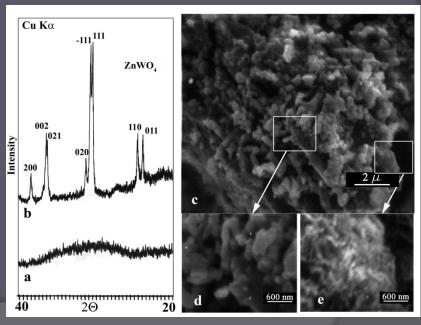
SEM micrographs of nanocrystalline powders of CaWO<sub>4</sub> at different magnification

Nanocrystalline powders of ZnWO<sub>4</sub>: a,b- XRD pattern powders synthesized at room temperature(a) and after annealing at 500°C ;

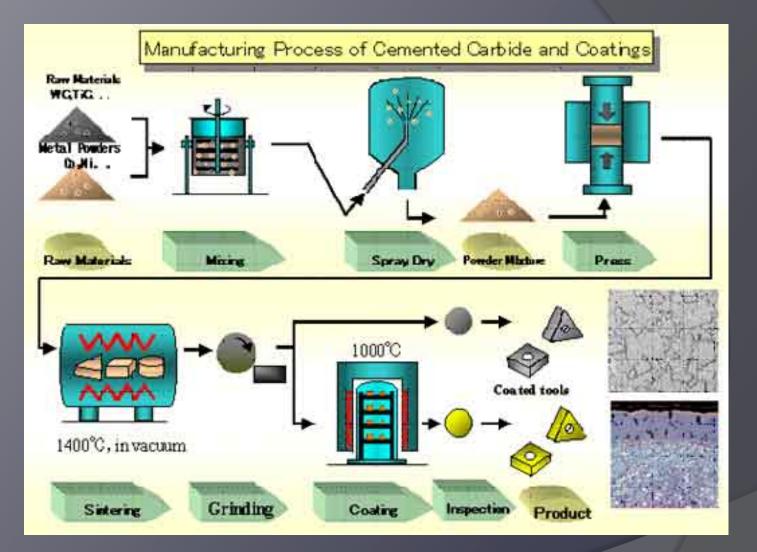
c,d,e - SEM micrographs at different magnification



SEM micrographs of nanocrystalline powders of CdWO<sub>4</sub> at different magnification

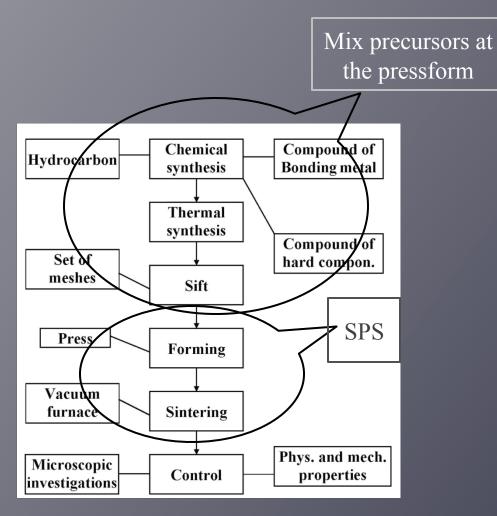


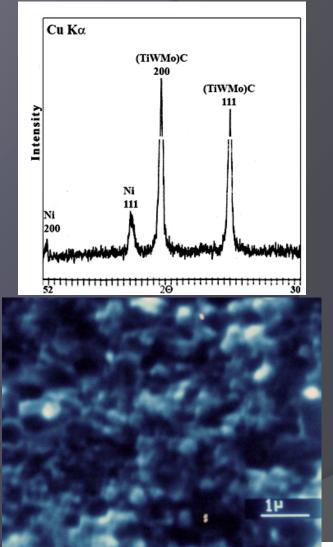
#### Hard Metals





#### Hard Metals





XRD pattern of (TiW,Mo)C-Ni system obtained from mixing precursors (TiH<sub>2</sub>, NiCl<sub>2</sub>, MoO<sub>3</sub>, WO<sub>3</sub>)



Hard Metals

 $\begin{array}{c} \text{TiH}_2 \longrightarrow \text{Ti+H}_2\\ \text{Ti+3TiCl}_4 \longrightarrow 4\text{TiCl}_3 \end{array}$ 

 $TiCl_3 + C_6H_6 \longrightarrow TiCl_3 \cdot C_6H_6$  $TiCl_3 + (CH_2)_6N_4 \longrightarrow Ti(CH_2)_6N_4Cl_3$ 

 $(NH_4)_2TiF_6 + NiCl_2 \cdot 6H_2O + 2(CH_2)_6N_4 \longrightarrow 2NH_4Cl + [Ni(H_2O)_6][TiF_6](C_6H_{12}N_4)_2$ 

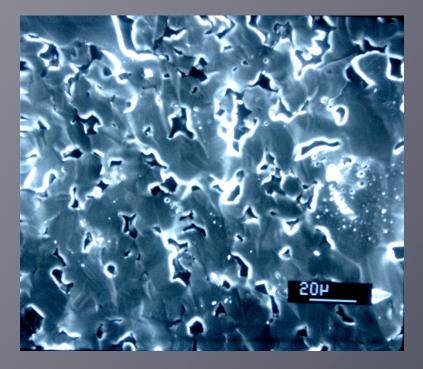
 $TiCl_4 + 4C_6H_5 NH_2 \longrightarrow NH_2 Ti(C_6H_5NH_2)_4Cl_4$  $TiCl_4 + 2NH_2(CH_2)_6NH_2 \longrightarrow Ti(2NH_2(CH_2)_6NH_2)_2Cl_2$ 

 $\begin{array}{rcl} \text{NiCl}_2 + 4\text{C}_6\text{H}_5\text{NH}_2 & \longrightarrow & [\text{Ni}(\text{C}_6\text{H}_5\text{NH}_2)_4]\text{Cl}_2 \\ \text{MoCl}_5 + 5\text{C}_6\text{H}_5\text{NH}_2 & \longrightarrow & [\text{Mo}(\text{C}_6\text{H}_5\text{NH}_2)_5]\text{Cl}_5 \\ \text{WCl}_6 + 6\text{C}_6\text{H}_5\text{NH}_2 & \longrightarrow & [\text{W}(\text{C}_6\text{H}_5\text{NH}_2)_6]\text{Cl}_6 \end{array}$ 

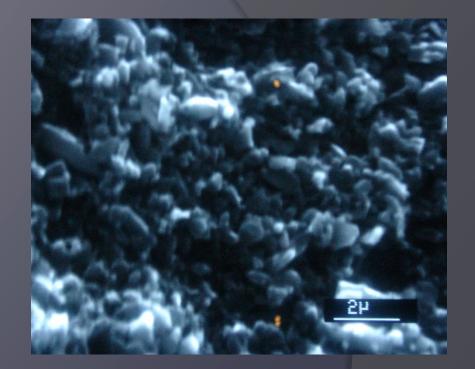
 $CHI_3 + MoCl_5 \longrightarrow [Cl_3Mo = CHI] + I_2 + Cl_2$  $CHI_3 + WCl_6 \longrightarrow [Cl_4W = CHI] + I_2 + Cl_2$ 



#### **Armor materials**



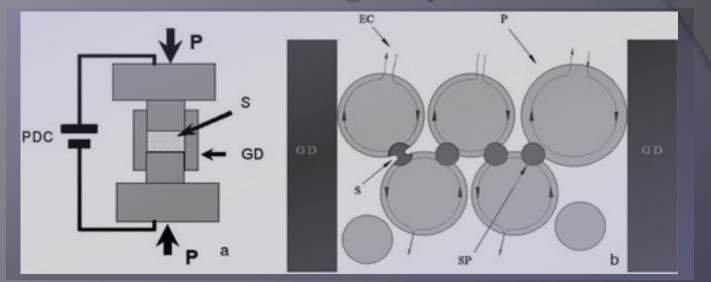
SEM images of BC standard armor material



SEM images of BC armor material sintered via SPS



#### SPS devise for sintering composition materials



Sintering temperature (max 2500°C) Current (max 5000 A) Applied pressure (max 200MPa)





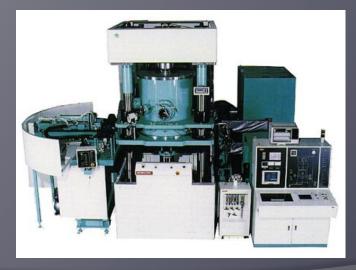
## Commercial SPS devises for sintering composition materials



SPS - 511S



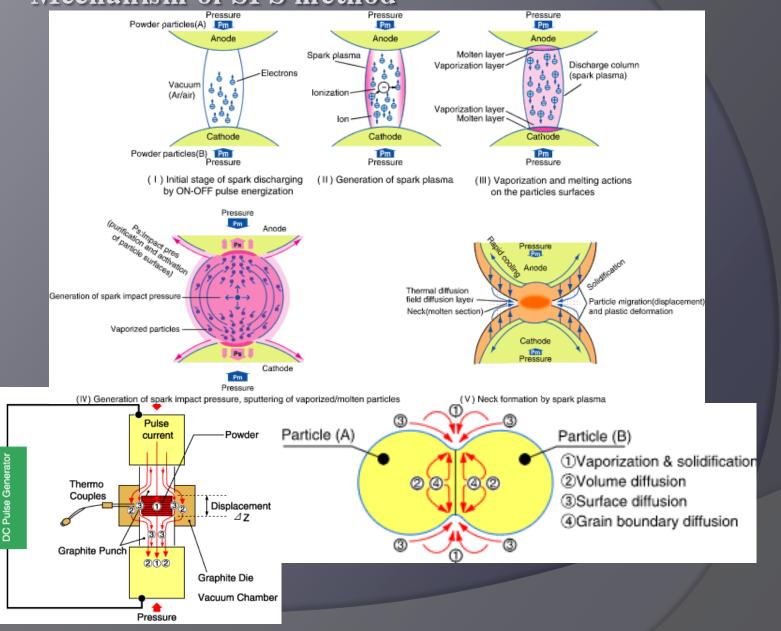
SPS - 2050





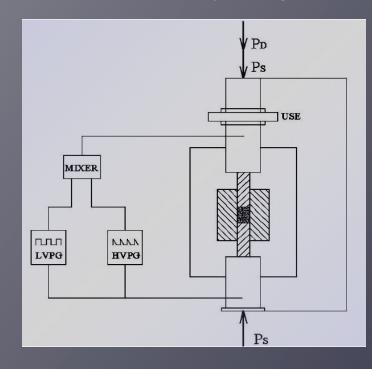
SPS-9.40

#### Mechanism of SPS method



RCSR

Schematic drawing of the modified SPS device that makes possible sintering non-conductive materials by using SPS method;



The device will be additionally equipped with *high voltage pulse generator*. Low voltage will be subjected to regulation and will be changed within the range of 10-20V, force of the current will be 3000-5000A. Magnitude of high voltage will range between 1-2 kv. Ultrasonic excitation device with 22-25 kHz frequency will be constructed for providing plasma processes and compaction assistance. The developed device will be equipped with the unit of *Pulsed dynamic loading*. Integral temperature of sintering will range within the **RCSR** interval 1000-1200°C.



## Thank you for attention

